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1 1 H 1.01 3 Li 6.94 11 Na 22.99 19 K 39.10 37 Rb 85.47 55 Cs 132.91 87 Fr (223) Atomic number Element Relative atomic mass 25 Mn 54.94 43 Tc 98.91 75 Re 186.21 61 Pm 146.92 93 Np (237)

CHEMISTRY STANDARD LEVEL PAPER 1

CHEMISTRY HIGHER LEVEL PAPER 1 INSTRUCTIONS TO CANDIDATES • Do not open this examination paper until instructed to do so. • Answer all the questions. • For each question, choose the answer you consider to be the best and indicate your choice on the answer sheet provided. • The periodic table is provided for reference on page 2 of this examination paper.

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Award [1] for each feature that is correct for both sodium and sodium chloride. Accept equation or half-equations for the reaction of sodium chloride in "reaction occurring" .

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CHEMISTRY HIGHER LEVEL PAPER 2 - St Leonard's College

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Chemistry Higher Level Non-Members - Topical Papers

There will be no whole paper penalty marks for accuracy AP, financial accuracy FP and units UP. Instead these skills will be assessed in particular questions and the marks applied according to the rules

MARKSCHEME

A. When vitamin D dissolves in fat, vitamin D is the solvent and fat is the solute. B. In a solution of NaCl in water, NaCl is the solute and water is the solvent. C. An aqueous solution consists of water dissolved in a solute. D. The concentration of a solution is the amount of solvent dissolved in 1 dm³ of solution.